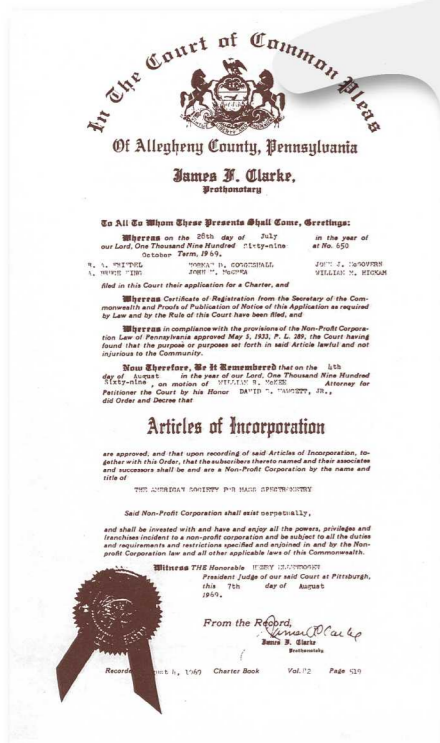


## ASMS Arranges Annual Conference

The Annual Conference was now "Arranged by the American Society for Mass Spectrometry in cooperation with ASTM Committee E-14." In July of 1969, The Articles of Incorporation of the American Society for Mass Spectrometry were filed in Allegheny County, Pennsylvania. The relationship between the new society and its ASTM progenitor was tenuous and in the process of forming. It was at this meeting that Harold Wiley was invited to regale the Conference banquet attendees with his recollections of the earliest origins of meetings on mass spectrometry; as well as some of the difficulties that early practitioners of the art faced. A transcript of his banquet speech is reproduced in a nearby poster.



By now, the Conference had moved well beyond its earlier emphasis on petrochemical applications of mass spectrometry and session titles included much broader areas of research such as:

- Studies of Negative Ions
- High Temperature Mass Spectrometry
- Mass Spectra of Biological Compounds
- Identification of Drugs and Drug Metabolites
- Pollution and Environment
- Mass Spectra of Solids
- Chemical Ionization
- Ion Molecule Reactions
- Computer Applications and Data Reduction

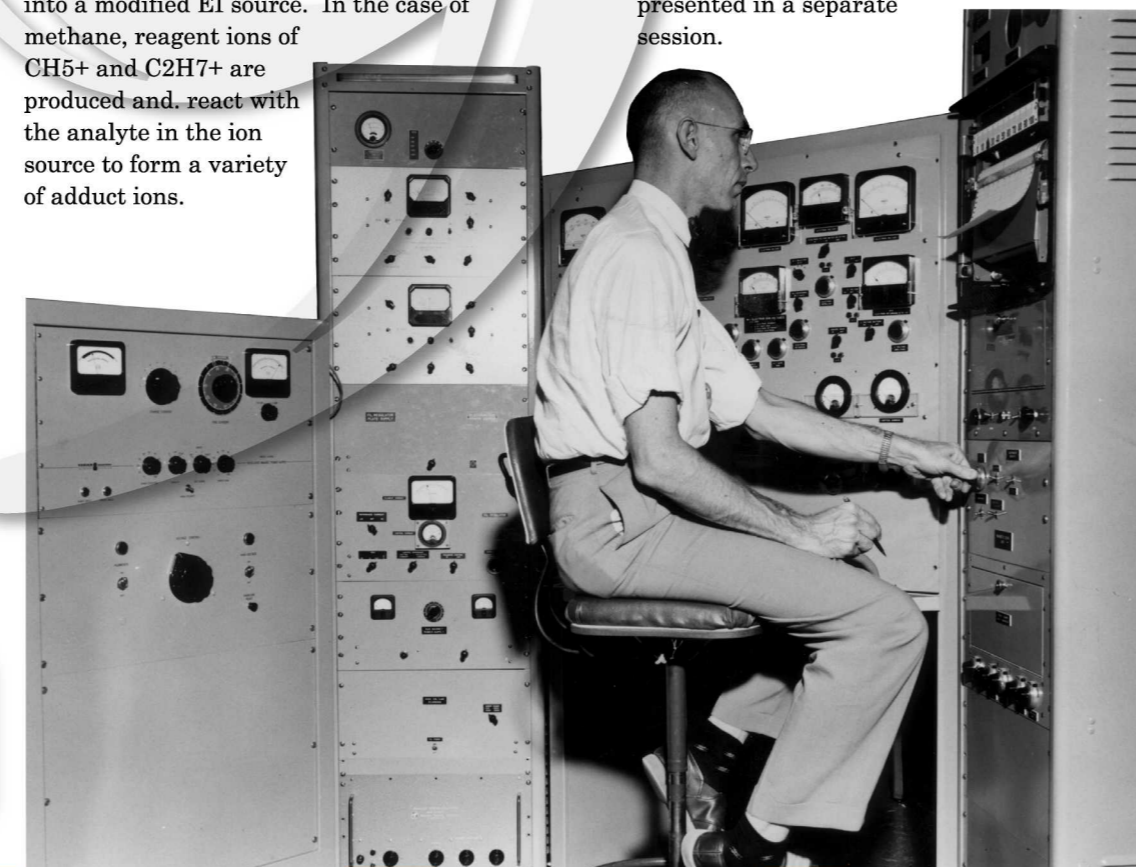
Over 200 hundred papers were presented in parallel oral sessions running the entire week. Attendees were excited to get to the meeting upon arrival in Dallas, but by the end of the week, they were just as anxious to get home and rest!

Frank Field at the console of the Humble Chemical Physics Mass Spectrometer ca 1960. Basic research with this instrument eventually led to the development of chemical ionization.

## A New Era in Ionization

In the early 60s, a group at the Research and Development Laboratory of the Humble Oil and Refining Company in Baytown Texas headed by Joe Franklin and Frank Field designed and built a mass spectrometer for the measurement of appearance and ionization potentials. It was later used to explore gaseous ionic reactions at increasingly higher pressures in research by Field and Burnaby Munson. An unexpected outcome of this high pressure research was the discovery and application of chemical ionization (CI). In CI, analyte ions are produced by gas phase reactions of the sample with reagent ions. The latter are produced by introducing a high pressure; close to a torr, of a gas, such as methane, into a modified EI source. In the case of methane, reagent ions of CH<sub>5</sub><sup>+</sup> and C<sub>2</sub>H<sub>7</sub><sup>+</sup> are produced and react with the analyte in the ion source to form a variety of adduct ions.

This development was significant because the only analytically useful method of ionization at the time was electron ionization. The latter technique had clear limitations since the sample had to be in the gas phase and the EI process imparted a large amount of energy to the analyte molecules. Thus, many non-volatile, polar compounds could not be analyzed by MS; and many labile compounds that could be ionized did not survive the ionization process to produce molecular ions. Chemical ionization was well suited to ionize these more 'difficult' compounds. By the time of the 1972 Annual Conference, a symposium on chemical ionization was organized and an additional dozen contributed papers were presented in a separate session.



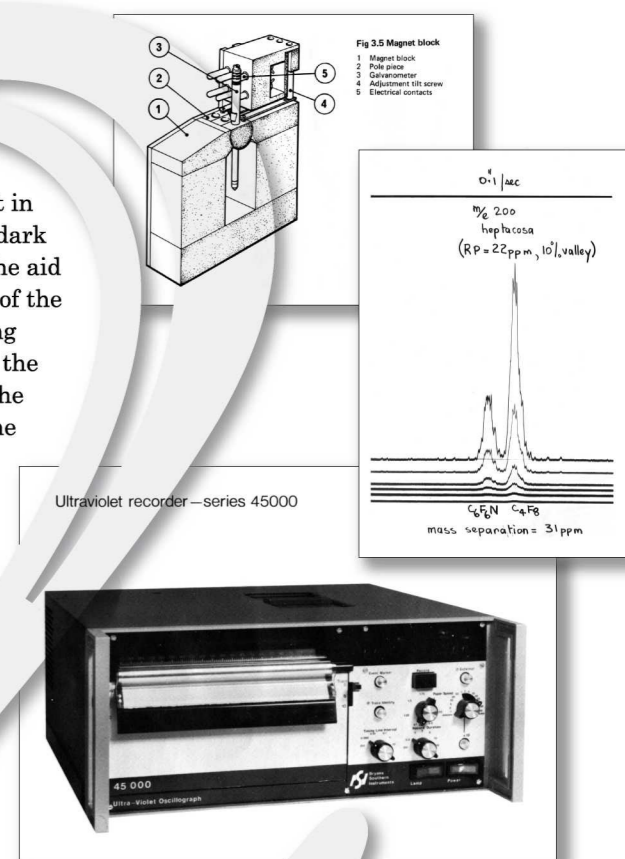
Modern day users operate the mass spectrometer from a computer terminal where they can examine mass spectra in real-time as they are acquired and can even change the operating mode of the mass spectrometer to perform different experiments based on the data in the most recent scan. Things were not always so easy.

For years, beginning with the sale of the first CEC instrument until well into the early 70's, the standard data acquisition device for mass spectrometry was the oscillographic recorder. This device was ideally suited for recording mass spectra, because it had the required frequency response and dynamic range.

The oscillographic recorder reflected a light beam from a mirror attached to a galvanometer onto photosensitive paper. Several such galvanometers could be placed in the recorder, each different gain. The photosensitive paper had to be kept in the dark room for developing, rinsing and fixing. This was accomplished with the aid of a baggy, black sleeve fitted over the arm up to the shoulder. The end of the sleeve had a light-tight flange which was manually engaged with a mating flange on the front of the recorder. Once the flange was secured, a door in the mating flange could be slid aside and the recording paper transferred into the sleeve. After twisting the end of the sleeve at the flange to keep out light, the door was closed, the flange disengaged and the recording paper transferred to the dark room without exposing it to light.

Developing the photographic paper took time and was messy. Since the chart paper was typically between 3 to 5 feet in length, the developing solutions were in long narrow tanks. When the developing process was complete, one had a length of wet chart paper that had to be hung out to dry. A significant advance in this technology came about when UV sensitive recording paper was introduced. This had the advantage that the exposed paper would 'develop' on exposure to ultraviolet light, eliminating the procedure of wet photo-developing. However, the spectra were not permanent. Stored in a dark file cabinet, they would last for a long time, but if left out on a desk for a day or two, the spectrum would essentially fade away.

After photo-processing and drying the spectrum, the masses had to be marked off and the intensities of pertinent peaks measured. The spectral marking routine was basically a counting exercise, the ions in the vicinity of water and air providing a recognizable starting point. Given the high dynamic range of the instrument, there was almost always a peak at every mass and simple counting could be performed. Measurement of peak intensities was somewhat more time consuming as there were typically five galvanometer traces at various attenuations; 1, 3, 10, 30, and 100 were commonly used. This accommodated a dynamic range in intensity of 1 in 10,000. Mass and intensity information was recorded on a large lined pad for latter analysis and computations. This laborious procedure was followed for every sample that was run and analyzed. There were no shortcuts, save drawing up a mass scale that could make the mass marking exercise a little quicker. Obviously it was a labor intensive exercise just to obtain a spectrum of mass and intensity.



MASS SPECTROMETRY	1963	1964	1965	1966	1967	1968	1969	1970	1971	1972
<b>HISTORY</b>	Soviet cosmonaut Valentina Tereshkova becomes the first woman to travel in space. Audiotape tapes are introduced. President John F. Kennedy is assassinated.	The Civil Rights Act bans many forms of institutional racism. Nikita Khrushchev is ousted from power in the Soviet Union in a hard-line coup led by Leonid Brezhnev. Murray Glen-Mann proposes that matter is made of quarks.	An electrical blackout grips the northeastern United States. Cosmic microwave radiation is discovered by Arno H. Penzias and Robert W. Wilson and is thought to be a residual effect of the big bang.	China's Cultural Revolution begins under the direction of Mao Zedong, throwing nearly all aspects of Chinese life into chaos. Miranda rules are established by the U.S. Supreme Court. Suharto comes to power in Indonesia.	Jocelyn Bell discovers pulsars. The Arab-Israeli Six-Day War ends. Israel occupies the West Bank, Golan Heights, Sinai Peninsula, and East Jerusalem. Christiaan Barnard performs the first successful human heart transplant on Louis Washansky. China explodes its first hydrogen bomb.	Martin Luther King, Jr., is assassinated. Robert F. Kennedy is assassinated. The Tet Offensive is launched by North Vietnamese forces.	The Cuyahoga River catches fire in Cleveland owing to the presence of high levels of flammable pollutants. The Woodstock festival takes place. Apollo 11 lands on the Moon.	The annual Conference on Mass Spectrometry and Allied Topics becomes a conference of the ASMS, initially in cooperation with the ASTM Committee E-14. Development of algorithms is begun for computer-based comparison of mass spectral data to libraries of known spectra for automated identification of unknown compounds. NASA researchers discover amino acids in carbonaceous chondrites. Eiji Osawa at Hokkaido University proposes a carbon compound with a three-dimensional "soccer ball" structure.	Glow discharge mass spectrometry is developed at IBM research laboratories in San Jose, California. "Static" secondary ion mass spectrometry is developed at the University of Cologne. GC-MS is used for clinical diagnosis of metabolic disorders. Reflection time-of-flight mass spectrometry is developed in Leningrad.	Democratic National Committee headquarters in the Watergate Hotel is burglarized. DDT is banned in the United States. The United States and the U.S.S.R. sign the Strategic Arms Limitation Treaty (SALT I).